

# Effect of Nb<sub>2</sub>O<sub>5</sub> on the aging and cytotoxicity of ZrO<sub>2</sub>-Y<sub>2</sub>O<sub>3</sub> based biomaterial

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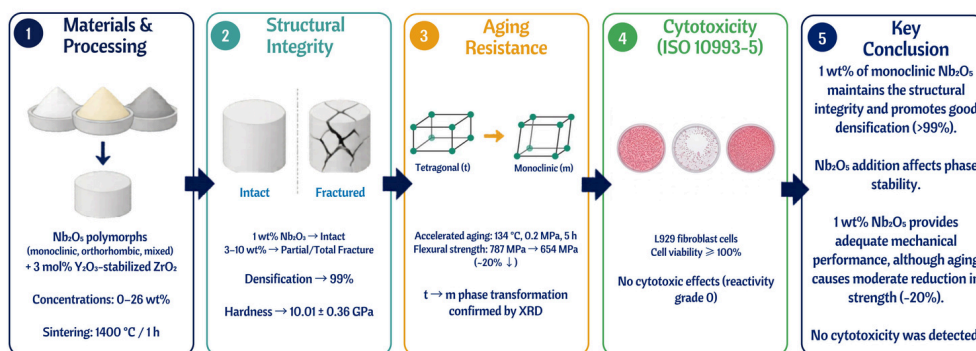
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## Effect of Nb<sub>2</sub>O<sub>5</sub> on Aging and Cytotoxicity of ZrO<sub>2</sub>-Y<sub>2</sub>O<sub>3</sub> Bioceramics

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**Abstract:** This study investigates the influence of different polymorphs and concentrations of niobium pentoxide (Nb<sub>2</sub>O<sub>5</sub>) on the aging resistance and cytotoxicity of Y<sub>2</sub>O<sub>3</sub>-stabilized zirconia (ZrO<sub>2</sub>) bioceramics. The goal was to evaluate the structural, mechanical, and biological responses of ZrO<sub>2</sub>-Y<sub>2</sub>O<sub>3</sub>-Nb<sub>2</sub>O<sub>5</sub> systems, addressing concerns related to low-temperature degradation (LTD) in biomedical applications. Different polymorphs of Nb<sub>2</sub>O<sub>5</sub> (monoclinic, orthorhombic, and mixtures) were incorporated at various concentrations (1–26 wt%) into 3 mol% Y<sub>2</sub>O<sub>3</sub>-stabilized ZrO<sub>2</sub> and processed through controlled sintering cycles. Phase compositions were characterized by X-ray diffraction and Raman spectroscopy. Mechanical and microstructural properties were evaluated through density measurements, Vickers microhardness, flexural strength testing, and SEM imaging. Accelerated aging tests simulated long-term in vivo conditions, and cytotoxicity was assessed via in vitro assays using L929 fibroblast cells in accordance with ISO 10993-5. Results demonstrated that the addition of 1 wt% monoclinic Nb<sub>2</sub>O<sub>5</sub> maintained structural integrity, mechanical strength, and biocompatibility, while higher concentrations led to cracking and tetragonal phase destabilization. Aging led to a moderate reduction in mechanical strength, and no cytotoxic effects were observed. The findings suggest that low-level monoclinic Nb<sub>2</sub>O<sub>5</sub> doping may enhance ZrO<sub>2</sub>'s viability for biomedical use.

**Keywords:** Biomaterials, ZrO<sub>2</sub>, Nb<sub>2</sub>O<sub>5</sub>, Aging, Cytotoxicity

## Introduction

It is well known that zirconia has intrinsic physical and chemical properties such as high hardness, wear resistance, low friction coefficient, and chemical inertness, which make it a material of interest for biomaterial applications [1, 2, 3]. It was only after the 1970s that there was an increase in zirconia applicability due to the understanding of the stabilization mechanism of the monoclinic (m) to tetragonal (t) ZrO<sub>2</sub> phase transformation and its toughening [4]. Studies have revealed that Y<sub>2</sub>O<sub>3</sub>, CeO<sub>2</sub>, CaO, and MgO doping is essential to retain the tetragonal or cubic phase of ZrO<sub>2</sub>, avoiding an increase in sample volume of approximately 3 to 4%. The most commonly used in medicine and dentistry is 2-3.5 mol% Y<sub>2</sub>O<sub>3</sub>-doped ZrO<sub>2</sub> (Y-TZP) [5]. The use of partially toughened zirconia as a biomaterial, especially as a femoral head, was successful from 1986 until about 2001. In 2001, reports indicated that 400 femoral implants failed within a short period, leading to concerns about the reliability of zirconia as an implant biomaterial [6-7]. The failure was attributed to accelerated aging of the metastable zirconia in an aqueous environment [3-8], known as low temperature degradation (LTD) which results in spontaneous and progressive t-m transformation thus compromising its mechanical properties. Alternatives have been proposed for improving the LTD resistance in the ZrO<sub>2</sub>-Y<sub>2</sub>O<sub>3</sub> system by adding secondary phase to it such as Al<sub>2</sub>O<sub>3</sub>, Nb<sub>2</sub>O<sub>5</sub>, Ta<sub>2</sub>O<sub>5</sub>, Gd<sub>2</sub>O<sub>3</sub>, Sm<sub>2</sub>O<sub>3</sub>, Yb<sub>2</sub>O<sub>3</sub>, and Sc<sub>2</sub>O<sub>3</sub> [9-11]. Regarding to Nb<sub>2</sub>O<sub>5</sub>, most of the published papers do not mention the polymorph used for the preparation of ZrO<sub>2</sub>-Y<sub>2</sub>O<sub>3</sub> ceramic [12-17]. Like zirconium oxide, niobium oxide undergoes a series of phase transformations depending on temperature. Although it is reported that Nb<sub>2</sub>O<sub>5</sub> exhibits different polymorphic crystalline phases such as orthorhombic, tetragonal, pseudohexagonal, and monoclinic as well as one amorphous phase (Nb<sub>2</sub>O<sub>5</sub>.nH<sub>2</sub>O), the most common are low-temperature orthorhombic

(< 1000 °C) and high-temperature monoclinic (≥ 1000 °C) forms [18-21]. When it comes to the use of ZrO<sub>2</sub>-Y<sub>2</sub>O<sub>3</sub> ceramics as biomaterials, knowledge of cytotoxicity is as important as the aging process. Thus, this work aims to analyze the impact of Nb<sub>2</sub>O<sub>5</sub> polymorphism and content on the aging behavior and cytotoxicity of the ZrO<sub>2</sub>-Y<sub>2</sub>O<sub>3</sub>-Nb<sub>2</sub>O<sub>5</sub> system.

## Material and Method

Nb<sub>2</sub>O<sub>5</sub> CBMM was used as received (a mixture of monoclinic and orthorhombic), Nb<sub>2</sub>O<sub>5</sub> CBMM treated at 1000 °C/3 h (monoclinic), Nb<sub>2</sub>O<sub>5</sub> Merck (orthorhombic), and TZ-3YSB-E Tosoh (3% Y<sub>2</sub>O<sub>3</sub>). Initially, the sintering of the mentioned niobium oxides was studied to verify whether the polymorphic transformation would cause any degradation of the specimens, since, like zirconia, phase transformations in niobium oxide lead to volumetric variations. Specimens were prepared by pressing at 400 MPa in 11 mm diameter molds and sintered at 1400 °C for 1 h.

Next, the effect of the type of niobium oxide on the sintering of TZ-3YSB-E was studied. In this case, mixtures were prepared with compositions presented in Table 1. The mixing process was carried out dry. The oxides were sieved using a 320-mesh sieve (TLM, Tamis) and mixed in different proportions in a ball mill (Yamato, UB 32) at approximately 400 rpm for about 3 h.

After milling, the mixtures were subjected to uniaxial pressing at 400 MPa for 10 seconds (VEB, Werkstoffprüfmaschinen Leipzig) in a mold with an approximate diameter of 1 cm. The green specimens were placed in a standard furnace (ME-1700) for a sintering cycle of 8 °C/min up to 600 °C, with a dwell time of 2 h to remove the binder from the ZrO<sub>2</sub>-Y<sub>2</sub>O<sub>3</sub>. After the binder removal period, the specimens were heated at a rate of 8 °C/min up to 800 °C, followed by 5 °C/min up to 1200 °C, and then at a heating rate of 3 °C/min from 1200 °C to 1400 °C, remaining at this temperature for 5 h.

**Table 1.** Studied Compositions.

Specimens	3Y-TZP – Tosoh					
	ZrO <sub>2</sub>		Y <sub>2</sub> O <sub>3</sub>		Nb <sub>2</sub> O <sub>5</sub> (*)	
	wt%	mol%	wt%	mol%	wt%	mol%
<b>0Nb-ZY</b>	94,80	97,00	5,20	3,00	0,00	0,00
<b>1Nb-ZY</b>	93,85	96,63	5,15	2,89	1,00	0,48
<b>3Nb-ZY</b>	91,96	95,69	5,04	2,86	3,00	1,44
<b>5Nb-ZY</b>	90,06	94,74	4,94	2,83	5,00	2,43
<b>7Nb-ZY</b>	88,16	93,75	4,84	2,81	7,00	3,34
<b>10Nb-ZY</b>	85,32	92,23	4,68	2,76	10,00	5,01
<b>26Nb-ZY</b>	74,00	85,72	0,0	0,0	26,0	14,28

(\*) Same concentration for all different polymorphs.

The composition that remained intact during the first stage was selected. As will be presented in the Results and Discussion section, this was composition 1Nb-ZY, as shown in Table 1. At this stage, the monoclinic Nb<sub>2</sub>O<sub>5</sub> CBMM was subjected to milling in a high-energy mill at 200 rpm for 5 h (Fritsch Pulverizette 5) to achieve a finer granulometry. This procedure was performed to obtain a system with greater homogeneity in the oxide mixture. In this context, the oxide mixing process was carried out dry, where the reagents were first sieved using a 320-mesh sieve, then mixed in well-defined portions using a ball mill rotating at approximately 400 rpm (YAMATO, UB 32) with alumina spheres for about 3 h. The mixtures were subjected to uniaxial pressing at 100 MPa for 10 s (EMIC, DL Line), followed by cold isostatic pressing at 250 MPa for 60 s (AIP, CP360). The green specimens were placed in a standard furnace (ME-1700-INTE) for a sintering cycle of 8 °C/min up to 600 °C, with a dwell time of 2 hours to eliminate organic elements. After the binder removal period, the specimens were heated at a rate of 8 °C/min up to 800 °C, followed by 5 °C/min up to 1200 °C, and then at a heating rate of 3 °C/min from 1200 °C to 1400 °C, remaining at this temperature for 1 hour.

## Characterizations

### X-Ray Diffraction and Raman Spectroscopy:

The raw materials and sintered specimens were characterized by X-ray diffraction (Shimadzu XRD 6000, using Co and Cu K $\alpha$  radiation) with a scanning rate of 1°/min in the range of 20° to 80°. For the 1Nb-ZY specimens, both before and after aging, specimens measuring 1 cm  $\times$  1 cm were used. Raman spectroscopy analyses were performed using a Witec 300-R confocal system with a 532 nm wavelength laser and a power of 1 mW.

**Scanning Electron Microscopy:** The raw materials and sintered specimens were analyzed by scanning electron microscopy (Quanta 400 FEG Zeiss) in secondary electron mode with an acceleration voltage of 20 kV.

**Flexural Strength:** The test was based on ASTM C1161 – 18 (Standard Test Method for Flexural Strength of Advanced Ceramics at Ambient Temperature) <sup>[22]</sup>. This method covers the determination of the bending strength of advanced ceramic materials at ambient temperature (Instron, 3369). In this type of test, the specimen is positioned appropriately and subjected to an increasing load at a constant speed until it fractures. In this test, 3 contact points were used, with a sample size of 10 specimens, a maximum distance between points of 40 mm, and a speed of 0.5 mm/min. The dimensions of the specimens followed those specified by the ASTM C1161 – 18.

**Apparent Density:** The apparent density measurements were carried out based on the method described in ASTM C 20 (Standard Test Methods for Apparent Porosity, Water Absorption, Apparent Specific Gravity, and Bulk Density of Burned Refractory Brick and Shapes by Boiling Water) [23]. In this context, the specimens were placed in an oven (Nova Ética) and then subjected to boiling distilled water using a heating plate (IKA, RH Basic 2) for 2 h. The dry masses, wet masses, and immersed masses were measured using an analytical balance adapted for the Archimedes method (Mettler Toledo, AB204-S/FACT).

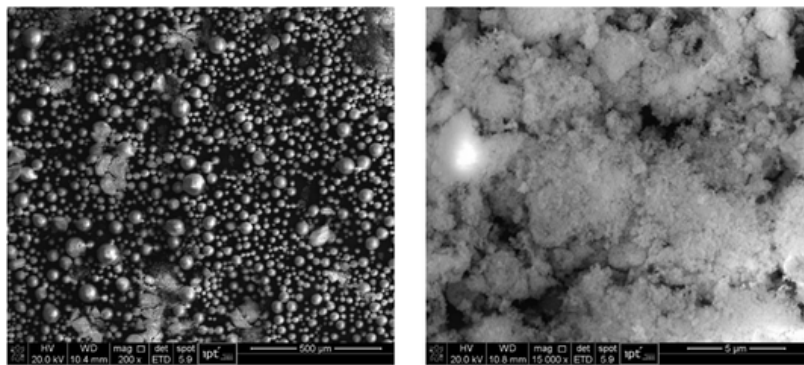
**Microhardness:** It was measured using standard equipment (EQUILAM, HVS-1000) with a force of 0.981 N applied in normal force for a duration of 15 s (SWAB, 2011). The analysis of the indentation dimensions was performed using the equipment's software. These measurements were conducted in accordance with ASTM C1327 – 15 (Standard Test Method for Vickers Indentation Hardness of Advanced Ceramics) [24].

**Accelerated Aging:** This test promotes accelerated aging of ceramic specimens by placing them in an autoclave (temperature of 134°C, pressure of 0.2 MPa for 5 h) (PHOENIX, AB25). This process standardizes that each hour in the autoclave is equivalent to 4 years in an in vivo system at a temperature of 37 °C [25].

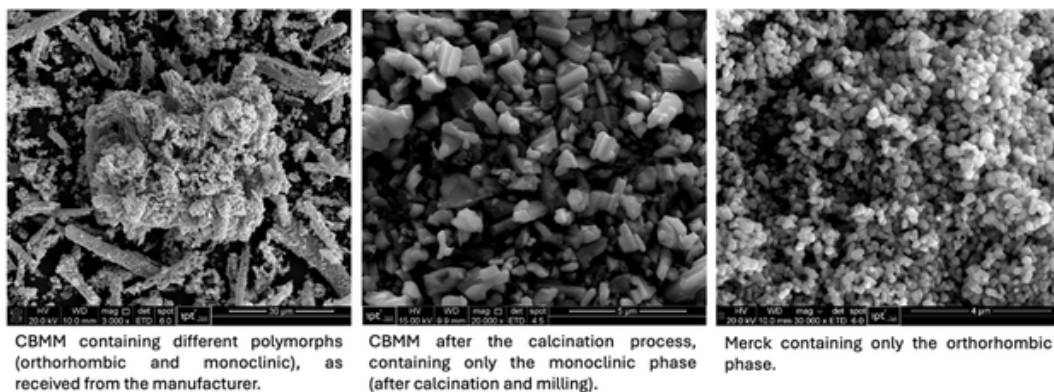
**In vitro cytotoxicity evaluations:** ISO 10993-5 (INTERNATIONAL ORGANIZATION for STANDARDIZATION. ISO 10993-5. Biological evaluation - Part 5: Test for in vitro cytotoxicity, 2009). In this analysis, indirect cytotoxicity on extracts and direct cytotoxicity on specimens were considered. Parameters: Cell line: L929 (ATCC No. CCL1, NCTC clone 929L); Culture Medium: DMEM (Sigma) containing 10% Bovine Fetal Serum (Sigma); Negative Cytotoxicity Control: filter paper discs; Positive Control of Cytotoxicity: latex glove discs [26].

## Results

**Characterization of the Raw Material:** The analysis using SEM showed powders with different particle size distributions, which can be observed in Figures 1, 2, and 3.



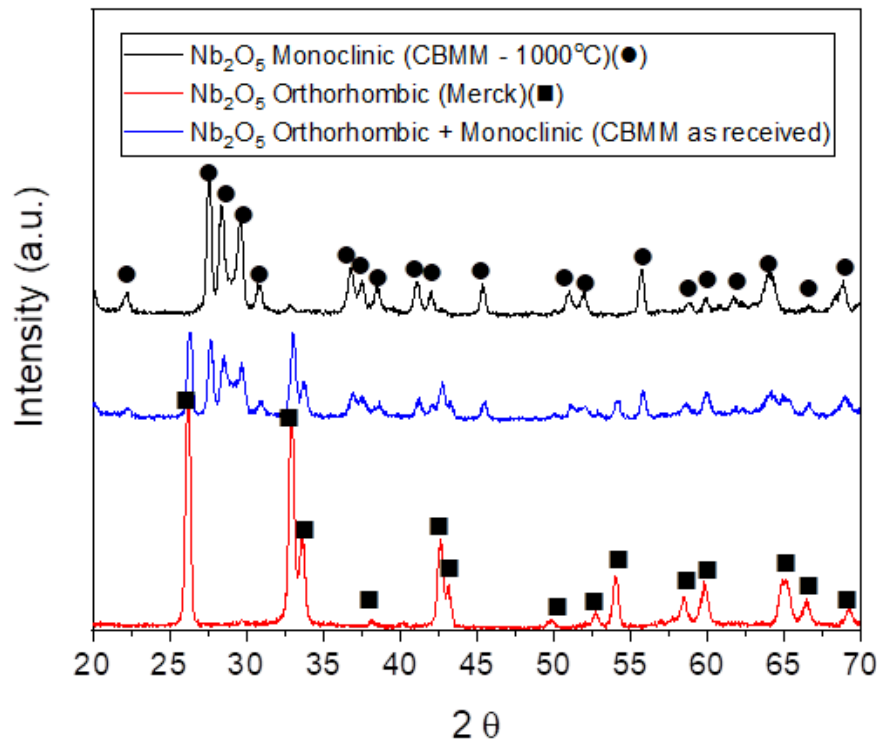
**Figure 1.** Powders from zirconium oxide manufactured by the company Tosoh and commercially named TZ-3YSB-E.



**Figure 2.** Powders from niobium oxide manufactured by CBMM and Merck and presented according to the type of polymorph present.

The characterization of the powders shown in Figure 2 was also performed by X-ray diffraction to understand not only the degree of purity of the

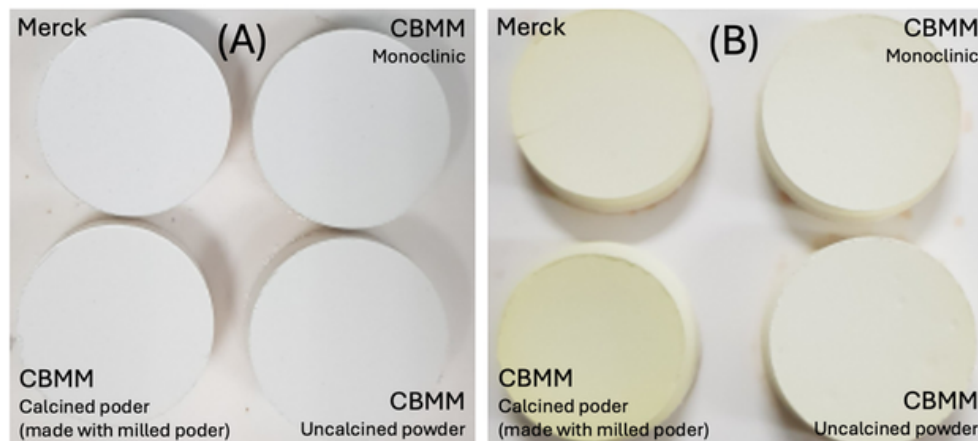
specimens but also to validate information provided by the manufacturers and existing literature.



**Figure 3.** X-ray diffraction of the niobium oxide powders.

**Visual analysis of the anisotropy coefficient of  $\text{Nb}_2\text{O}_5$ :** Manning et al. reported the thermal coefficient anisotropy of monoclinic  $\text{Nb}_2\text{O}_5$  [18]. According to these authors, the difference in the expansion coefficient in the **a** ( $5.3 \times 10^{-6}/^\circ\text{C}$ ), **b** (0)

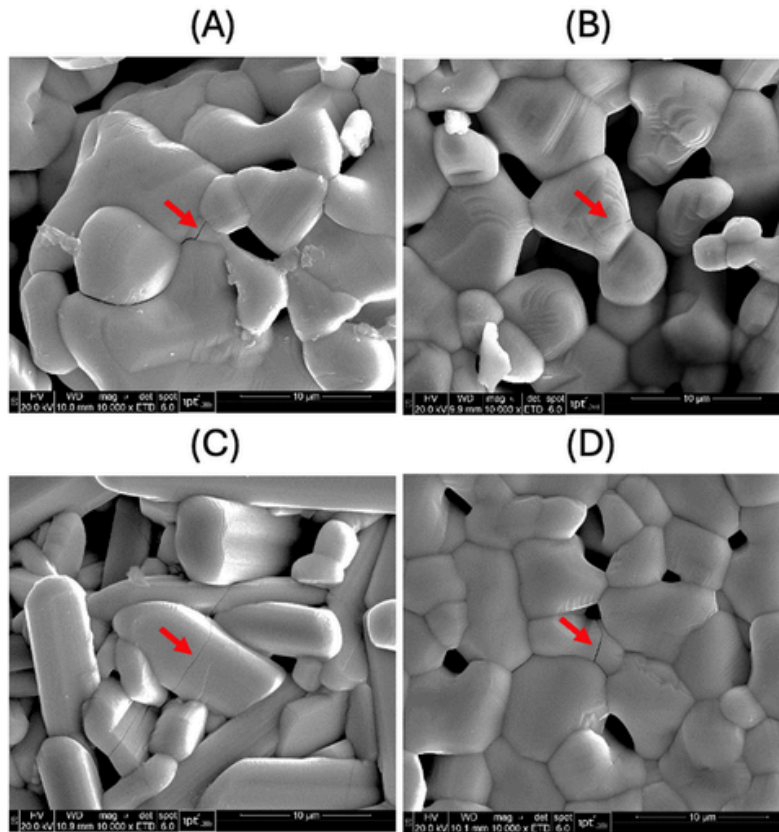
and **c** ( $5.9 \times 10^{-6}/^\circ\text{C}$ ) of the crystal structure of monoclinic niobium oxide leads to the formation of microcracks in the sintered body of this material.



**Figure 4.** (A) Green  $\text{Nb}_2\text{O}_5$  pellets and (B) pellets sintered at  $1400^\circ\text{C}/1\text{h}$ .

The groups of sintered specimens, showed in Figure 4, were subjected to SEM analysis in order to investigate the presence of microcracks. These

observations were conducted on the flat surface of the specimens as presented in Figure 5.



**Figure 5.** SEM micrographs of the Nb<sub>2</sub>O<sub>5</sub> pellets sintered at a maximum temperature of 1400 °C/1 h. (A) surface of specimens from monoclinic and orthorhombic polymorphs (B) only monoclinic (C) only monoclinic pulverized after calcination and (D) only orthorhombic. Source: Developed by the author.

The formation of microcracks without the fracture of the specimens is possible through the understanding of the phenomenon common to its class as a polycrystalline ceramic material. During the cooling of Nb<sub>2</sub>O<sub>5</sub> from the sintering temperature, unequal contraction of the crystals occurs, generating internal stresses. These stresses are more pronounced in directions with high thermal expansion coefficients (directions a and c) and compressive in directions with low coefficients (direction b). When these stresses exceed the strength of the crystals or the bonds between them, internal fracture occurs, leading to the formation of microcracks<sup>[18]</sup>.

The presence of microcracks causes anomalous thermal expansion, or even contraction, at lower temperatures. Initially, the individual thermal expansion of the crystals is accommodated by the cracks, preventing the body from expanding as a whole. As the temperature increases, the fracture surfaces begin to recombine, resulting in the expansion of the body<sup>[18, 27-28]</sup>.

The hysteresis observed during the heating and cooling cycles of sintered specimens can be attributed to the occurrence and recombination of microcracks. Therefore, an inversely proportional relationship between the density of microcracks and porosity is observed, a characteristic that may explain the maintenance of the macrostructure of the developed specimens, even in the presence of microcracks. These specimens were developed with grain sizes and compaction profiles of the green specimens in such a way as to be less susceptible to catastrophic fractures<sup>[18]</sup>.

**Developing specimens based on ZrO<sub>2</sub>-Y<sub>2</sub>O<sub>3</sub>-Nb<sub>2</sub>O<sub>5</sub> system:** Based on the understanding of the sintering behavior of Nb<sub>2</sub>O<sub>5</sub>, the process of forming specimens containing ZrO<sub>2</sub> (manufacturer TOSOH with product described as TZ-3YSB-E) and Nb<sub>2</sub>O<sub>5</sub>, with different polymorph configurations, was initiated. Table 2 presents the developed profiles and the identification of catastrophic failures after the sintering process.

**Table 2** - Relationship between the type of Nb<sub>2</sub>O<sub>5</sub> polymorph, polymorph content, and integrity of sintered specimens of the TZ-3YSB-E-Nb<sub>2</sub>O<sub>5</sub> system after cooling.

Nb <sub>2</sub> O <sub>5</sub> Polymorph Type	Mass Fraction of Nb <sub>2</sub> O <sub>5</sub> and ZrO <sub>2</sub> (Tosoh) [wt%]	Fracture Profile	Nomenclature
<b>Monoclinic</b>	1-99	<b>Intact</b>	TZ-3YSB-E _1M
	3-97	Partial Fracture	
	5-95	Partial Fracture	
	7-93	Total Fracture	
	10-90	Total Fracture	
	26-74	<b>Intact</b>	
<b>Polymorph Mixture</b>	1-99	<b>Intact</b>	TZ-3YSB-E _1MO
	3-97	Partial Fracture	
	5-95	Total Fracture	
	7-93	Total Fracture	
	10-90	Total Fracture	
<b>Orthorhombic</b>	1-99	<b>Intact</b>	TZ-3YSB-E _1O
	3-97	Partial Fracture	
	5-95	Total Fracture	
	7-93	Total Fracture	
	10-90	Total Fracture	

Among the sintered specimens that remained intact, those containing 1% and 26% Nb<sub>2</sub>O<sub>5</sub> were identified. In contrast, the remaining specimens exhibited partial or complete fractures immediately after the proposed thermal cycle. Studies on the 26Nb<sub>2</sub>O<sub>5</sub>-74ZrO<sub>2</sub> [wt%] system suggest the formation of the Zr<sub>6</sub>Nb<sub>2</sub>O<sub>17</sub> [28-29]. The differences in integrity levels can be observed in Figure 6.

Under appropriate conditions, the formation of a specific crystalline structure of Zr<sub>6</sub>Nb<sub>2</sub>O<sub>17</sub> occurs instead of a mixture of individual oxides. This structure exhibits an orthorhombic arrangement, allowing zirconium and niobium atoms to bond stably through oxygen linkages, forming a crystalline lattice. Additionally, the interaction between niobium and zirconium atoms, facilitated by Nb-O-Zr bonds, is fundamental to the material's stability. These bonds promote a cohesive three-dimensional network, preventing the separation of components into distinct phases [28, 30].

In other words, the structure of Zr<sub>6</sub>Nb<sub>2</sub>O<sub>17</sub> is described as a superstructure composed of metal and oxygen subcells, which accommodate different concentrations of Nb. This structural accommodation capability contributes to the material's stability. Finally, when stoichiometric molar ratios are used, X-ray diffraction analysis reveals the formation of a single Zr<sub>6</sub>Nb<sub>2</sub>O<sub>17</sub> phase, with no evidence of other oxides [31-33].

From the perspective of ionic interactions, the addition of Nb<sub>2</sub>O<sub>5</sub> to yttria-stabilized zirconia modifies the densification mechanisms primarily through the generation of Nb<sup>5+</sup>→Zr<sup>4+</sup> substitutional defects, which increase the concentration of oxygen vacancies and enhance the anionic diffusion responsible for the initial stages of sintering. This effect may be further reinforced by the formation of small amounts of eutectic phases or transient liquids that lower the onset temperature of densification and accelerate pore closure [34-36]. Furthermore, Nb<sup>5+</sup> doping can promote grain-boundary solute drag, supporting high densification while limiting excessive grain growth [37].

Regarding phase stability, Nb<sup>5+</sup> tends to destabilize the tetragonal phase of zirconia because the oxygen vacancies generated by pentavalent doping do not stabilize tetragonality as effectively as those associated with Y<sup>3+</sup>. Increased ionic disorder and lattice distortion decrease the free energy of the tetragonal phase, thereby facilitating the martensitic tetragonal→monoclinic transformation during cooling or under mechanical stress [1, 38-39]. Thermodynamically, the higher defect density reduces the energetic barrier for this transformation, making the material more susceptible to phase change [40].

Thus, Nb<sub>2</sub>O<sub>5</sub> exerts a dual influence on the ZrO<sub>2</sub>-Y<sub>2</sub>O<sub>3</sub> system: it accelerates densification by increasing diffusional mobility during sintering, while simultaneously reducing tetragonal stability by altering defect equilibria

and structural symmetry. These phenomena directly affect the final microstructure and the mechanical properties of zirconia-based biomedical ceramics<sup>[41,42]</sup>.



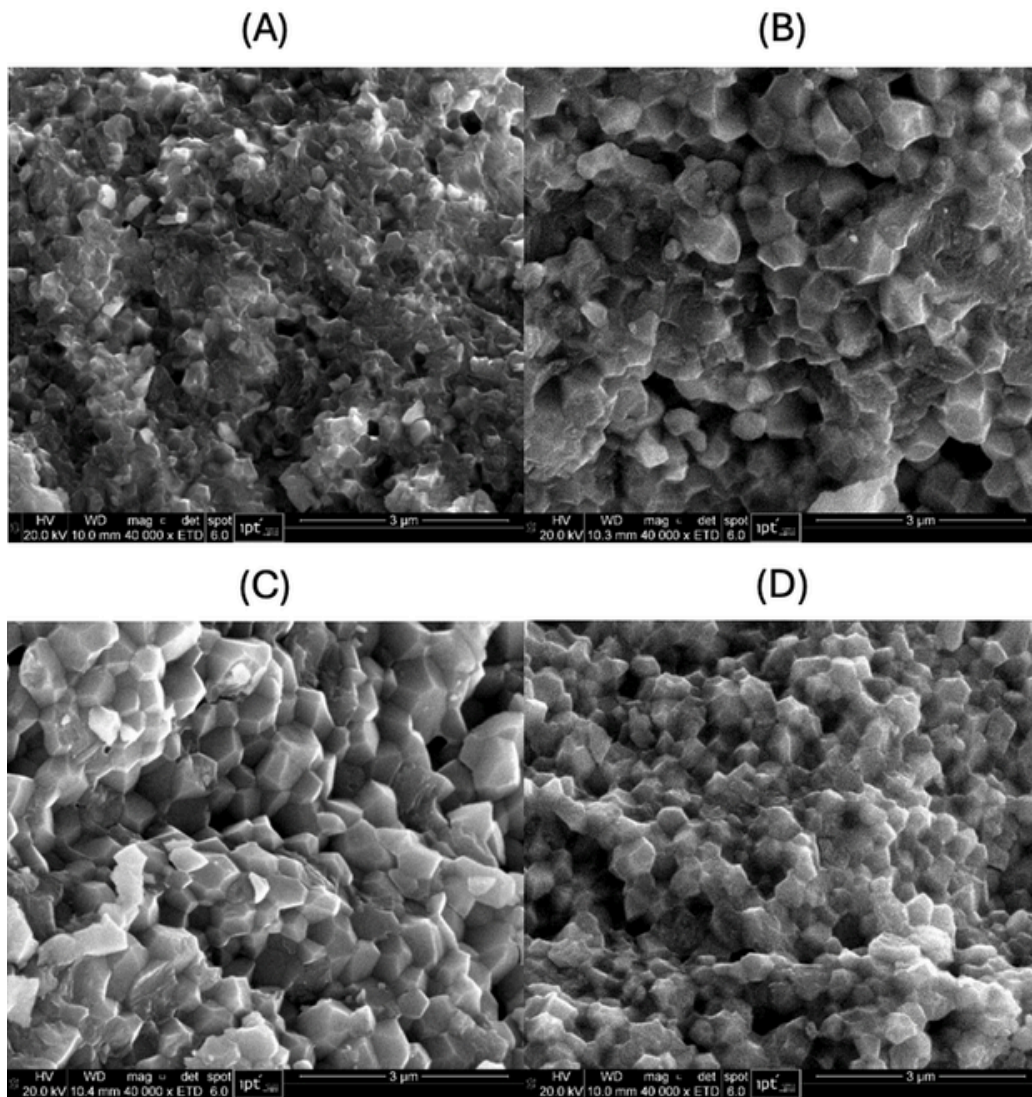
**Figure 6** - Photographs of sintered ceramic specimens based on their integrity levels (examples from the sintered samples). (A) Intact (TZ-3YSB-E\_1M). (B) Partial fracture (3-97 [%wt] with monoclinic Nb<sub>2</sub>O<sub>5</sub>). (C) Total fracture (10-90 [%wt] with monoclinic Nb<sub>2</sub>O<sub>5</sub>). Sintered from green specimens diameter: 11 mm.

**Densification:** Based on the experimental results obtained, it is understood that the experimental density was  $6.04 \pm 0.01$  g/cm<sup>3</sup>, a value statistically identical to the theoretical one. This high degree of densification suggests that, even with the potential formation of microcracks due to thermal mismatch associated with Nb<sub>2</sub>O<sub>5</sub>-rich residual regions, part of the oxide distributed within the system may have reacted with the zirconia matrix or formed secondary phases, thereby reducing the extent of microcracking during cooling. Because such possible secondary compounds often exhibit more isotropic thermal behavior, their presence may contribute to maintaining the high densification observed. Furthermore, the resulting grain size, mixing methodology, and pressing conditions may have collectively favored the consolidation of specimens achieving approximately 99% of the theoretical density.

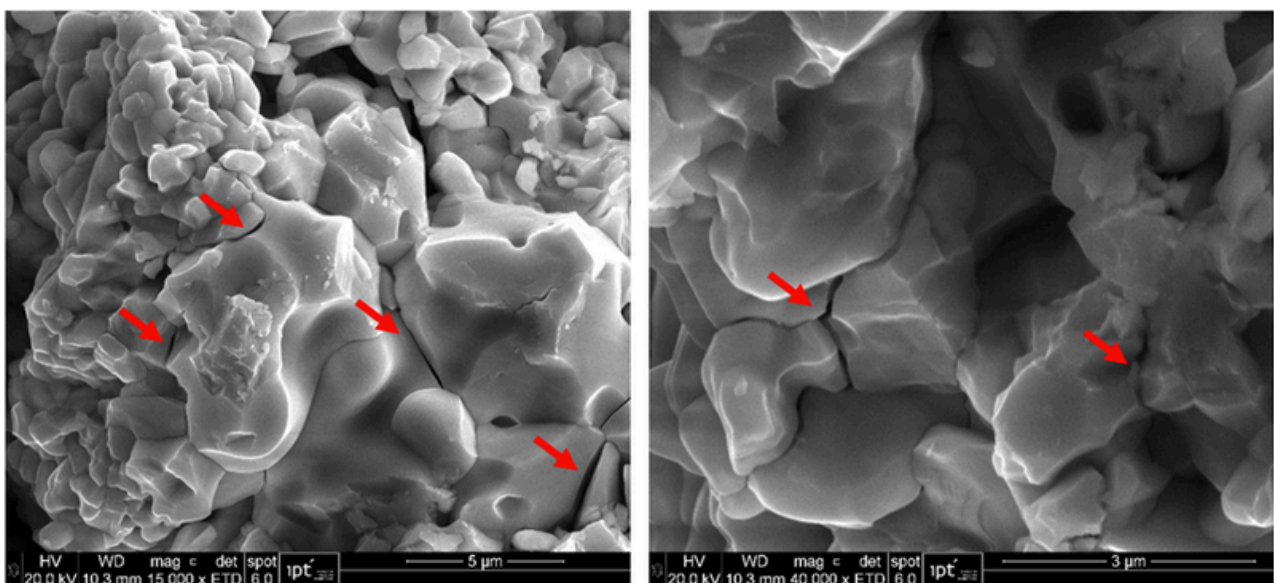
This densification mechanism can be rationalized by the defect chemistry introduced by Nb<sup>5+</sup> into the zirconia lattice. When incorporated substitutionally into Zr<sup>4+</sup> sites, Nb<sup>5+</sup>, whose ionic radius (~0.64–0.69 Å) is slightly smaller than that of Zr<sup>4+</sup> (0.72 Å), induces local lattice distortions and charge-compensating defects, such as cation vacancies, which can modify species mobility during sintering. These defect structures may enhance mass transport

at specific temperatures and promote pore elimination, thus aiding densification in Nb<sub>2</sub>O<sub>5</sub>-doped zirconia systems. Additionally, niobium redistribution along grain boundaries can alter grain growth kinetics and diffusion pathways, contributing to the sintering behavior reported in the literature for similar ZrO<sub>2</sub>-Nb<sub>2</sub>O<sub>5</sub>-Y<sub>2</sub>O<sub>3</sub> ceramic systems<sup>[43-44]</sup>.

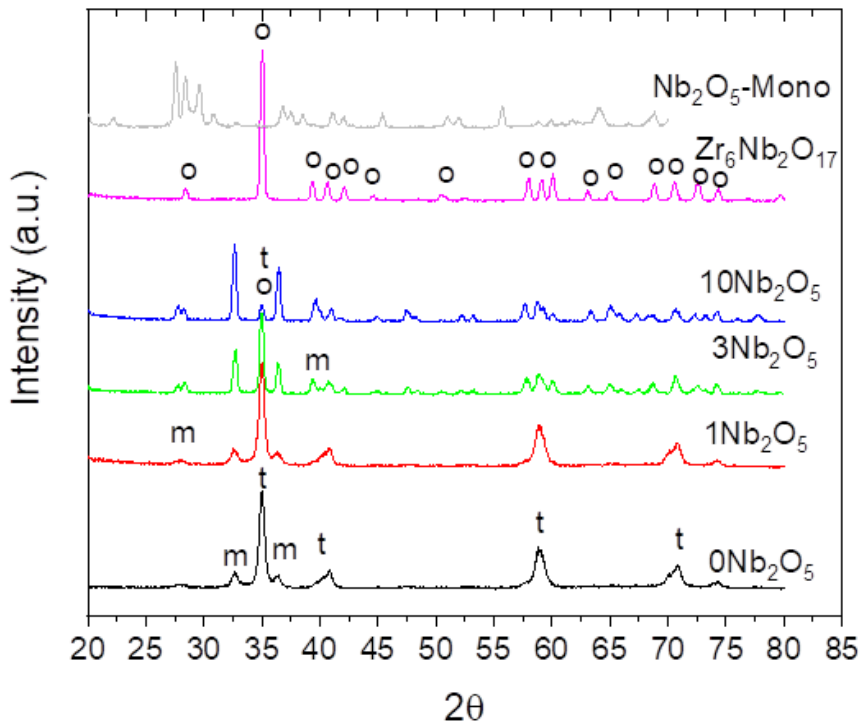
**Microstructure:** Figure 7 presents scanning electron microscope (SEM) images of the TZ-3YSB-E sintered specimens containing 1% of Nb<sub>2</sub>O<sub>5</sub> with different types of polymorphs. The microstructure is finer for the compositions without Nb<sub>2</sub>O<sub>5</sub> (0% addition) and with 1% Nb<sub>2</sub>O<sub>5</sub> from Merck, attributed to their finer granulometry compared to that of the Nb<sub>2</sub>O<sub>5</sub> from CBMM. Figure 8 displays micrographs of the fractured surface of TZ-3YSB-E sintered specimens containing 10% of monoclinic Nb<sub>2</sub>O<sub>5</sub>, highlighting the cracks that were developed during cooling due to the destabilization of partially stabilized ZrO<sub>2</sub>-Y<sub>2</sub>O<sub>3</sub>. The X-ray diffraction results shown in Figure 9 indicate that the tetragonal phase decreases as the concentration of Nb<sub>2</sub>O<sub>5</sub> increases to 10 wt% leading to destabilization of partially stabilized zirconia. However, when the ZrO<sub>2</sub>:Nb<sub>2</sub>O<sub>5</sub> ratio is set at 6:1, the compact is intact due to the formation of Zr<sub>6</sub>Nb<sub>2</sub>O<sub>17</sub>.



**Figure 7** - Fracture surface of specimens of (A) TZ-3YSB-E, (B) TZ-3YSB-E with 1%Nb<sub>2</sub>O<sub>5</sub> Monoclinic phase (CBMM), (C) TZ-3YSB-E with 1%Nb<sub>2</sub>O<sub>5</sub> Monoclinic and Orthorhombic phases (CBMM), and (D) TZ-3YSB-E with 1%Nb<sub>2</sub>O<sub>5</sub> Orthorhombic phase (Merck).



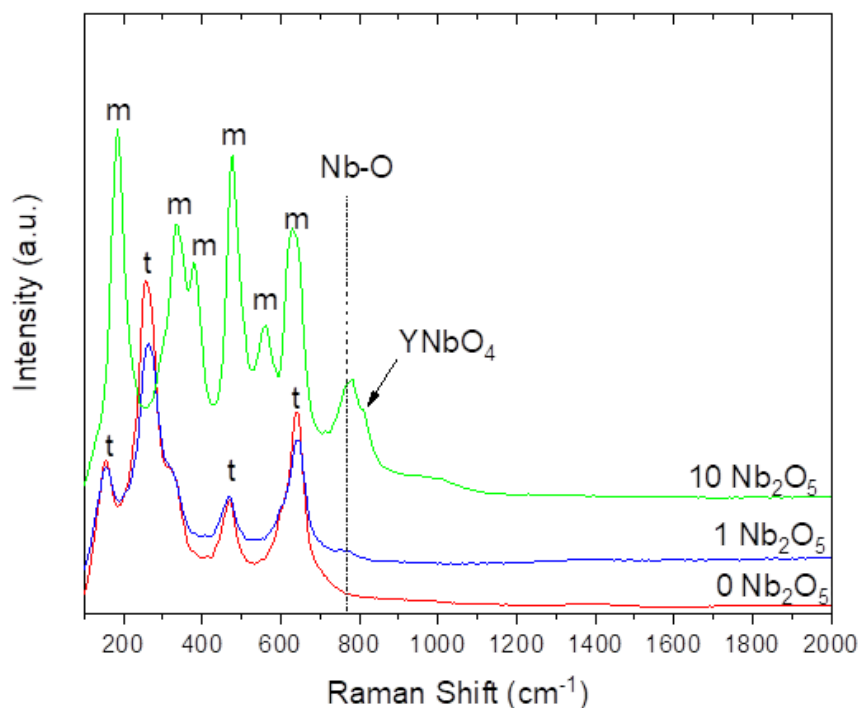
**Figure 8** - SEM micrographs of the fracture surface of pellets composed by 90 wt% TZ-3YSB-E and 10 wt% Nb<sub>2</sub>O<sub>5</sub> monoclinic phase that was sintered at 1400 °C/1h.



**Figure 9** -X-ray diffractograms of pellets with different systems of TZ-3YSB-E and monoclinic Nb<sub>2</sub>O<sub>5</sub> (CBMM). The concentrations where 0 wt% Nb<sub>2</sub>O<sub>5</sub> (0Nb<sub>2</sub>O<sub>5</sub>), 1 wt% Nb<sub>2</sub>O<sub>5</sub> (1Nb<sub>2</sub>O<sub>5</sub>), 3 wt% Nb<sub>2</sub>O<sub>5</sub> (3Nb<sub>2</sub>O<sub>5</sub>), 10 wt% Nb<sub>2</sub>O<sub>5</sub> (10Nb<sub>2</sub>O<sub>5</sub>), 26 wt% Nb<sub>2</sub>O<sub>5</sub> (Zr<sub>6</sub>Nb<sub>2</sub>O<sub>17</sub>), and only monoclinic Nb<sub>2</sub>O<sub>5</sub>. Radiation of Co K $\alpha$ .

Figure 10 shows the Raman spectra of compacts prepared with 0.0, 1.0, and 10.0 wt% of Nb<sub>2</sub>O<sub>5</sub> from CBMM (monoclinic). In the specimen containing 10 wt% Nb<sub>2</sub>O<sub>5</sub> (equivalent to 5.01 mol%), the tetragonal phase was not observed, indicating a complete destabilization of the zirconia tetragonal phase. Although the

band at 811 cm<sup>-1</sup> is quite weak, it suggests the presence of YNbO<sub>4</sub>, as reported by Kim et al. [45]. The band located around 780 cm<sup>-1</sup> is attributed to the Nb-O bond within the ZrO<sub>2</sub>-Y<sub>2</sub>O<sub>3</sub>-Nb<sub>2</sub>O<sub>5</sub> system, as noted by Lee, Jang, and Kim and Tan et al. [21, 46].



**Figure 10** - Raman spectra of pellets with different systems of TZ-3YSB-E and monoclinic Nb<sub>2</sub>O<sub>5</sub> (CBMM). The concentrations where 0 wt% Nb<sub>2</sub>O<sub>5</sub> (0Nb<sub>2</sub>O<sub>5</sub>), 1 wt% Nb<sub>2</sub>O<sub>5</sub> (1Nb<sub>2</sub>O<sub>5</sub>), and 10 wt% Nb<sub>2</sub>O<sub>5</sub> (10Nb<sub>2</sub>O<sub>5</sub>).

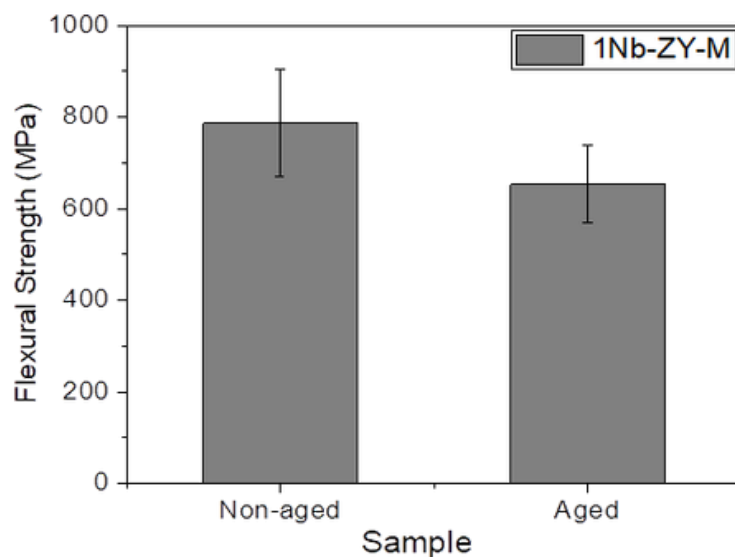
**Microhardness:** The Vickers microhardness value measured for sintered specimens composed by TZ-3YSB-E with 1 wt% of monoclinic Nb<sub>2</sub>O<sub>5</sub> (CBMM) resulting in approximately (10.010 ± 0.355) GPa. Although these values are lower than those of commercial Y-TZP, they were not found to be significantly different in practice, falling within 70.30% to 77.72% of the performance reported in the literature [47-48]. In addition to these findings, a deeper examination of the statistical significance of the microhardness results is essential to properly contextualize their mechanical relevance. Although the measured mean value and relatively narrow standard deviation suggest consistent experimental response, reliance solely on the absence of statistical differences may mask subtle but functionally important variations, especially in advanced ceramics where microstructural heterogeneity strongly influences hardness behavior. In zirconia-based systems, factors such as grain boundary chemistry, dopant distribution, and residual stress fields can generate local hardness fluctuations that are not captured when only P-values are considered [49-51]. Complementary metrics, such as effect size, confidence intervals, and intra-sample variability analyses, provide a more comprehensive framework for determining whether the observed differences could influence wear resistance or long-term reliability, even when they fall below conventional thresholds of statistical significance. Previous studies have shown that small changes in hardness, though statistically non-significant, may meaningfully alter mechanical response under cyclic loading or contact fatigue in biomedical applications [52]. Incorporating these broader descriptors therefore enhances the interpretation of the microhardness findings and aligns the analysis with best practices in ceramic materials characterization.

**Accelerated Aging:** As illustrated in Figure 11, there was an average reduction of about 20% in the strength of the aged (654.19 ± 83.52 MPa) compared to the non-aged specimens (787.16 ± 116.53 MPa). This reduction can be attributed to the transformation of tetragonal to monoclinic zirconia. Figure 12 indicates that the intensity of the peak at 28.15° corresponding to monoclinic ZrO<sub>2</sub> is more pronounced in the compact aged at 134 °C for 5 h. This finding suggests that the addition of 1 mol% Nb<sub>2</sub>O<sub>5</sub> (from CBMM - monoclinic) influences the transformability of zirconium oxide in our study. Lee et al. [9] investigated the effects of adding Nb<sub>2</sub>O<sub>5</sub> and Y<sub>2</sub>O<sub>3</sub> on the aging of ZrO<sub>2</sub>. They found that the m→t phase transformation did not occur in a composition containing 90.24 mol% ZrO<sub>2</sub>, 5.31 mol% Y<sub>2</sub>O<sub>3</sub>, and 4.75 mol% Nb<sub>2</sub>O<sub>5</sub>, even when aged in the temperature range of 200 °C to 400 °C for up to 1000 h. The authors suggest that the retention of low-temperature degradation (LTD) rates is due to the balancing of charge neutrality, facilitated by the replacement of Nb<sup>5+</sup> and Y<sup>3+</sup> ions for Zr<sup>4+</sup> ions. The control of stabilization and LTD in zirconium oxide, which is partially or fully stabilized by Y<sub>2</sub>O<sub>3</sub>, relies on several factors. These include grain size, tetragonality (c/a), the amount of additive (in this case, Nb<sup>5+</sup>), charge balancing, vacancy annihilation, and the formation of NO<sub>4</sub>, YO<sub>8</sub>, and ZrO<sub>8</sub> units. These factors are critical in managing the internal stresses responsible for the m-t transformation [9, 13-14, 17, 53]. In our study, given the molar ratios used, the selected raw materials, and the resulting microstructure, we found that equilibrium conditions were not achieved.

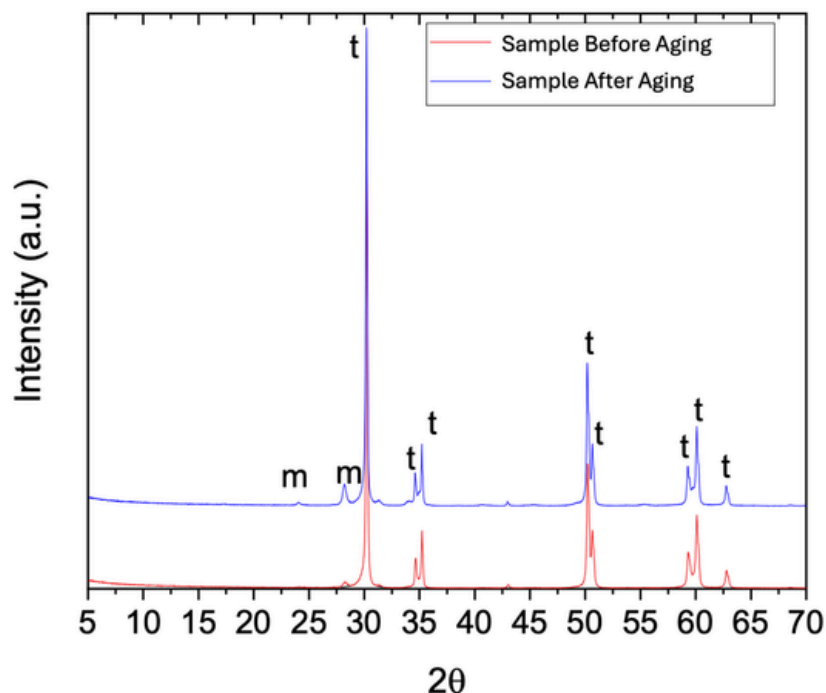
In addition to these observations, a more detailed examination of the statistical significance of the flexural strength results is essential for a complete interpretation of the material's mechanical behavior. Although the average reduction of approximately 20% in flexural strength between non-aged and aged specimens appears substantial, the relatively high standard deviations associated with both

conditions (116.53 MPa and 83.52 MPa, respectively) indicate considerable data dispersion, which may reduce the statistical power of conventional hypothesis-testing. Ceramics such as Y-TZP and Nb<sub>2</sub>O<sub>5</sub>-doped zirconia often exhibit variability arising from microstructural heterogeneities, including flaw population, grain size distribution, and local phase assemblages, that can influence fracture behavior without necessarily producing statistically significant differences when evaluated solely by P-values [54–56]. Complemen-

tary analyses such as effect size, Weibull modulus evaluation, confidence intervals, and probability-of-failure modeling provide a deeper understanding of whether the observed reductions in strength reflect true mechanical degradation or fall within the variability expected for polycrystalline ceramic systems [57]. Incorporating these statistical descriptors would therefore enhance the robustness of the interpretation and allow the mechanical implications of aging and phase transformation to be more accurately contextualized.



**Figure 11** - Flexural strength of the sintered specimens composed by TZ-3YSB-E with 1 wt% of monoclinic Nb<sub>2</sub>O<sub>5</sub> from CBMM (1Nb-ZY-M) before and after aging.

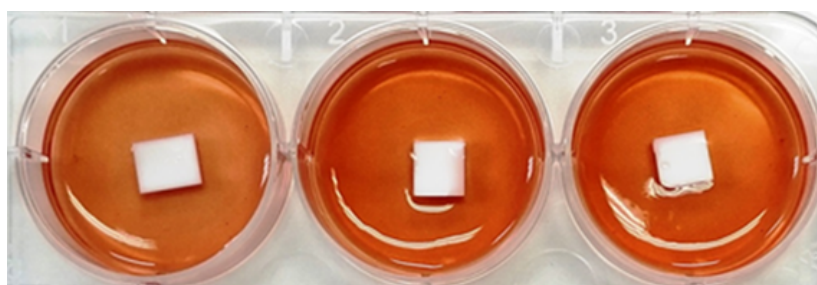


**Figure 12** - X-ray diffractograms of the sintered specimens composed by TZ-3YSB-E with 1 wt% of monoclinic Nb<sub>2</sub>O<sub>5</sub> from CBMM before and after aging.

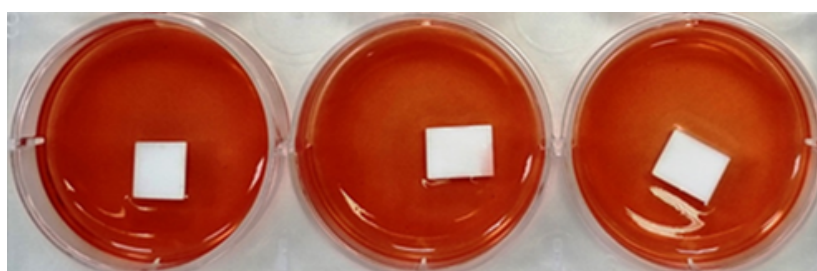
**Cytotoxicity:** Although zirconia-based ceramics are generally considered bioinert, there are limited studies on the  $ZrO_2$ - $Y_2O_3$ - $Nb_2O_5$  system concerning this important material property. Pradham et al.<sup>[58]</sup> investigated cell viability using L-929 fibroblast cell lines for implants coated with  $Nb_2O_5$  and  $TiO_2$ . Their findings indicated that cell viability was higher for the  $TiO_2$ -coated samples (122%) compared to the growing cell population. In contrast, the implants coated with  $Nb_2O_5$  showed variable viability, ranging from 64% to 105%, which was partly dependent on the atomic structure of the materials. Specifically, the authors found that hexagonal  $Nb_2O_5$  demonstrated bioactivity, whereas amorphous and orthorhombic forms did not. However, the study did not include an analysis of monoclinic  $Nb_2O_5$ <sup>[58-63]</sup>. Obata et al.<sup>[64]</sup> demonstrated that exposure of osteoblastic cells to  $Nb^{5+}$  ions released from niobium-containing bioactive materials did not compromise cell viability or morphology; on the contrary, the authors reported maintenance of metabolic activity and even increased osteogenic markers, such as alkaline phosphatase activity and matrix mineralization. Additionally, although Dsouki et al.<sup>[65]</sup> primarily evaluated systemic biological res-

ponses in vivo, their findings also support the notion of low biological reactivity and the absence of adverse cellular effects attributable to niobium-based compounds. Collectively, these observations reinforce that  $Nb_2O_5$ , even when incorporated into zirconia-based ceramics, is unlikely to induce cytotoxic responses under typical extract test conditions. Such results further support the relevance of cytotoxicity analysis for  $ZrO_2$ - $Y_2O_3$ - $Nb_2O_5$  systems.

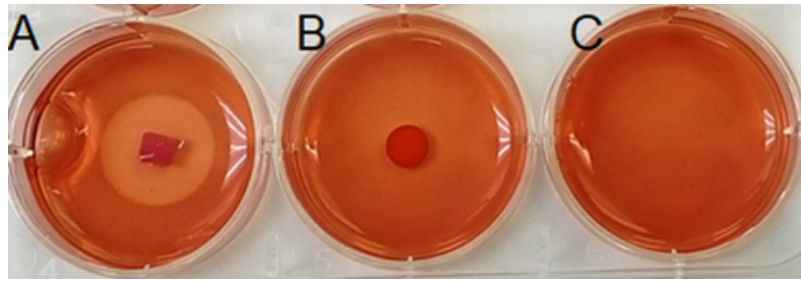
In this study, it is observed the cell monolayer both macroscopically and microscopically after 24 hours of exposure to the test materials. The control (TZ-3YSB-E) and specimens made by TZ-3YSB-E with 1 wt% of monoclinic  $Nb_2O_5$  from CBMM showed a reactivity degree of 0 ( $n = 3$ ), indicating that there were no alterations in the cells compared to the negative control. The validation of the test was confirmed by the presence of an inhibition halo and cellular changes around the latex sample, which served as a positive control for cytotoxicity. These results are illustrated in Figures 13 and 14. Additionally, Figure 15 displays the positive control (A), the negative control (B), and a well without any interference (C).



**Figure 13** - Presentation of the wells containing the control specimen (TZ-3YSB-E). L929 cell line culture submerged in the agar layer containing approximately 1 cm<sup>2</sup> of specimen placed above.



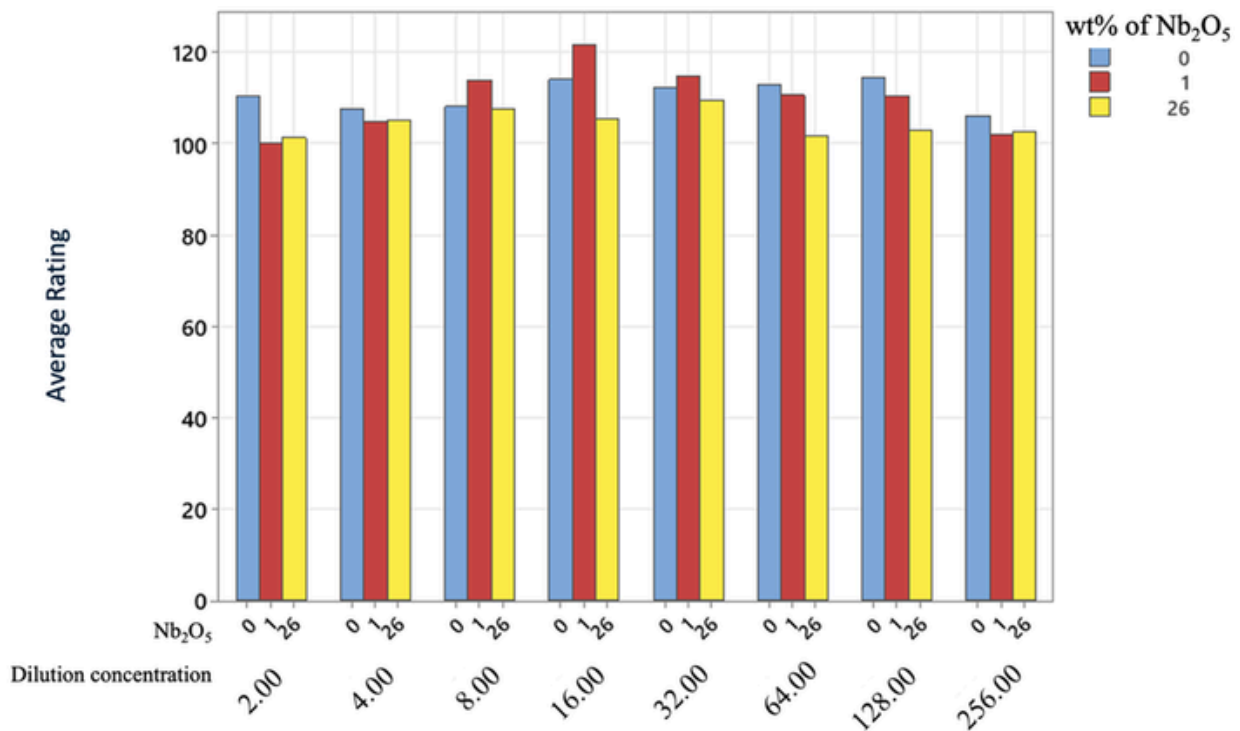
**Figure 14** - Presentation of the wells containing the sintered specimens composed by TZ-3YSB-E with 1 wt% of monoclinic  $Nb_2O_5$  from CBMM. L929 cell line culture submerged in the agar layer containing approximately 1 cm<sup>2</sup> of specimens allocated above.



**Figure 15** - L929 cell line culture submerged in an agar layer containing approximately 1 cm<sup>2</sup> of sample placed on top. A = positive/cytotoxic control. B = negative/non-cytotoxic control. C = well containing cells without interaction.

The quantitative test results, obtained by performing serial dilutions in base 2 of the extracts from the sintered specimens composed by TZ-3YSB-E with 0, 1 and 26 wt% of monoclinic Nb<sub>2</sub>O<sub>5</sub> from CBMM indicated no cytotoxicity towards the reference cells used in

the experiment, as shown in Figure 16. All conditions demonstrated a cell viability of 100% or higher. The P-value for the analysis of variance was 0.151, which is greater than 0.05 and suggests that there is no statistical evidence to indicate that the means are different.



**Figure 16** - Analysis of variations in cell viability percentages of the sintered specimens composed by TZ-3YSB-E with 0 (0), 1 (1) and 26 (26) wt% of monoclinic Nb<sub>2</sub>O<sub>5</sub> from CBMM control samples as a function of increasing dilution at level 2.

## Conclusions

According to the study on the addition of different polymorphs in the ZrO<sub>2</sub>-Y<sub>2</sub>O<sub>3</sub> system with 3 mol% concentration, and considering the molar ratios of Nb<sub>2</sub>O<sub>5</sub>/Y<sub>2</sub>O<sub>3</sub>, the chosen raw materials, and the procedural conditions, the following conclusions can be drawn:

- The Nb<sub>2</sub>O<sub>5</sub> compacts exhibited microcracks in their microstructure after heat treatment and cooling from 1400 °C to 25 °C,
- The addition of up to 1% by weight of Nb<sub>2</sub>O<sub>5</sub> resulted in intact compacts, regardless of the Nb<sub>2</sub>O<sub>5</sub> polymorph used,

- The addition of 3% to 10% by weight led to fractured compacts after heat treatment and cooling, independent of the Nb<sub>2</sub>O<sub>5</sub> polymorph used in the formulations,
- The densification of the compacts was higher than 99% at 1400 °C /1 h,
- The hardness of the ZrO<sub>2</sub>-Y<sub>2</sub>O<sub>3</sub> system with 1% of monoclinic Nb<sub>2</sub>O<sub>5</sub> was (10.010 ± 0.355) GPa,
- The flexural strength of the sintered ZrO<sub>2</sub>-Y<sub>2</sub>O<sub>3</sub> compact with 1% monoclinic Nb<sub>2</sub>O<sub>5</sub> decreased from (787.16 ± 116.53) MPa to (654.19 ± 83.52) MPa after aging. This reduction indicates a transformation from tetragonal to monoclinic (t→m), as observed through X-ray analyses,
- The cell viability and proliferation studies indicated that none of the specimens exhibited cytotoxicity following the addition of monoclinic Nb<sub>2</sub>O<sub>5</sub> to ZrO<sub>2</sub>-Y<sub>2</sub>O<sub>3</sub>.

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